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(μ -2,2'-Biquinoliny-4,4'-dicarboxylato- κ^2 O:O')bis[(dimethylformamide- κ O)-triphenyltin(IV)]

Li-Bin Wang

Department of Chemistry, Tonghua Teachers' College, Tonghua, Jilin 134001, People's Republic of China

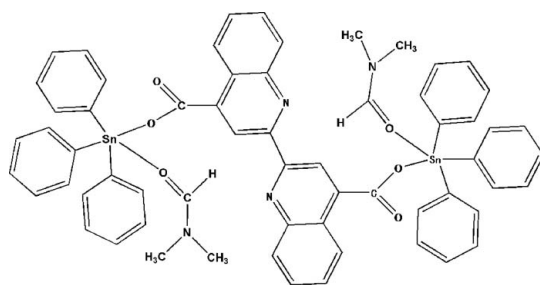
Correspondence e-mail: wanglibin0612@yahoo.com.cn

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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.010$ Å; R factor = 0.054; wR factor = 0.170; data-to-parameter ratio = 18.6.

The molecule of the title compound, $[\text{Sn}_2(\text{C}_6\text{H}_5)_6(\text{C}_{10}\text{H}_5\text{NO}_2)_2(\text{C}_3\text{H}_7\text{NO})_2]$, is centrosymmetric. Two triphenyltin groups are bridged by a 2,2'-biquinoliny-4,4'-dicarboxylate dianion (L) through its two carboxylate groups. Each Sn atom displays a distorted trigonal-bipyramidal geometry composed of three phenyl groups, one carboxylate O atom from the L dianion and one O atom of N,N -dimethylformamide. The molecules display aromatic π - π stacking (face-to-face distance 3.406 Å).

Related literature

For related literature, see: Basu Baul *et al.* (2004).

Experimental

Crystal data

 $[\text{Sn}_2(\text{C}_6\text{H}_5)_6(\text{C}_{10}\text{H}_5\text{NO}_2)_2(\text{C}_3\text{H}_7\text{NO})_2]$
 $M_r = 1188.47$
Monoclinic, $C2/c$
 $a = 30.4592$ (17) Å
 $b = 9.5344$ (5) Å
 $c = 22.0343$ (14) Å
 $\beta = 123.051$ (2)°
 $V = 5363.5$ (5) Å³
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.99$ mm⁻¹
 $T = 293$ (2) K
 $0.54 \times 0.48 \times 0.39$ mm

Data collection

Bruker APEX CCD area-detector diffractometer
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.590$, $T_{\max} = 0.678$

15947 measured reflections
6203 independent reflections
4618 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.033$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.054$
 $wR(F^2) = 0.170$
 $S = 1.08$
6203 reflections

334 parameters
H-atom parameters constrained
 $\Delta\rho_{\max} = 2.82$ e Å⁻³
 $\Delta\rho_{\min} = -1.14$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

Sn1—C13	2.125 (4)	Sn1—C1	2.127 (4)
Sn1—C7	2.127 (4)	Sn1—O1	2.187 (4)
C13—Sn1—C7	119.75 (18)	C13—Sn1—O1	88.78 (16)
C13—Sn1—C1	114.85 (16)	C7—Sn1—O1	103.41 (17)
C7—Sn1—C1	119.16 (19)	C1—Sn1—O1	102.48 (19)

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINTE* (Bruker, 1999); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL-Plus* (Sheldrick, 1990); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WW2083).

References

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supplementary materials

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(μ -2,2'-Biquinolinyl-4,4'-dicarboxylato- κ^2 O:O')bis[(dimethylformamide- κ O)triphenyltin(IV)]

L.-B. Wang

Comment

To further widen the scope of application of organotin compounds, there is a need to prepare new series of organotin compounds. In this paper, the structure of (I) is described.

The title compound (I) contains one neutral $[\text{Sn}_2(\text{C}_6\text{H}_5)_6(\text{C}_9\text{H}_5\text{NCOO})_2]$ molecule and two *N,N*-dimethyl formamide solvent molecules. As shown in Fig. 1, two triphenyltin groups are bridged by the 2,2'-biquinoline-4,4'-dicarboxylate dianion, *L*, through its two carboxylate groups. The central Sn atom adopts distorted SnC_3O_2 trigonal bipyramid geometry composed of three phenyl groups, one carboxylate O atom from *L* and one O atom of *N,N*-dimethyl formamide. The *N,N*-dimethyl formamide, included as solvent molecule, forms no hydrogen-bonding interactions with any other atoms in the structure. In addition, the bond between Sn1 and O3 (2.605 (3) Å) is much more weaker than that of Sn1 and O1 (2.187 (4) Å) (Basu Baul *et al.*, 2004). The *L* ligand of each molecule provides two quinoline rings to form slipped π - π interaction with the face-to-face distance (*d*) being about 3.406 Å (Fig. 2).

Experimental

A mixture of Na_2L (0.388 g, 1 mmol) and Ph_3SnCl (0.770 g, 2 mmol) in 25 ml ethanol was refluxed for 12 h and then filtered. The resulting precipitates were washed with ethanol. Crystals suitable for X-ray diffraction were obtained by slow evaporation of a *N,N*-dimethyl formamide solution.

Refinement

All H-atoms bound to carbon were refined using a riding model with $d(\text{C}-\text{H}) = 0.93$ Å, $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ for aromatic and 0.96 Å, $U_{\text{iso}} = 1.5U_{\text{eq}}(\text{C})$ for CH_3 atoms. The minimum electron-density peak is located 0.96 Å from atom Sn1.

Figures

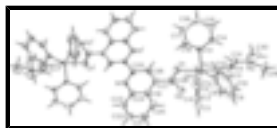


Fig. 1. View of the compound (I). Displacement ellipsoids are drawn at the 30% probability level. (symmetry code: (i) $1 - x, 1 - y, 1 - z$)

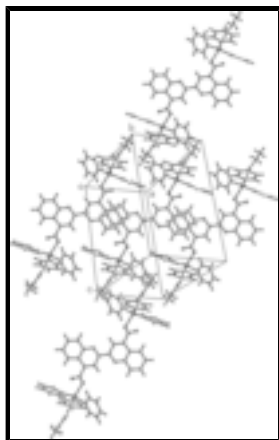


Fig. 2. A packing diagram of (I).

(μ -2,2'-Biquinoliny-4,4'-dicarboxylato- κ^2 O:O')bis[(dimethylformamide- κ O)triphenyltin(IV)]

Crystal data

[Sn₂(C₆H₅)₆(C₁₀H₅NO₂)₂(C₃H₇NO)₂]

$M_r = 1188.47$

Monoclinic, $C2/c$

Hall symbol: $-C\ 2yc$

$a = 30.4592\ (17)\ \text{\AA}$

$b = 9.5344\ (5)\ \text{\AA}$

$c = 22.0343\ (14)\ \text{\AA}$

$\beta = 123.051\ (2)^\circ$

$V = 5363.5\ (5)\ \text{\AA}^3$

$Z = 4$

$F_{000} = 2408$

$D_x = 1.472\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation

$\lambda = 0.71073\ \text{\AA}$

Cell parameters from 6203 reflections

$\theta = 1.6\text{--}28.3^\circ$

$\mu = 0.99\ \text{mm}^{-1}$

$T = 293\ (2)\ \text{K}$

Block, colourless

$0.54 \times 0.48 \times 0.39\ \text{mm}$

Data collection

Bruker APEX CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 293(2)\ \text{K}$

ω scans

Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)

$T_{\min} = 0.590$, $T_{\max} = 0.678$

15947 measured reflections

6203 independent reflections

4618 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.033$

$\theta_{\max} = 28.3^\circ$

$\theta_{\min} = 1.6^\circ$

$h = -40 \rightarrow 26$

$k = -12 \rightarrow 10$

$l = -26 \rightarrow 29$

Refinement

Refinement on F^2

Least-squares matrix: full

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

$R[F^2 > 2\sigma(F^2)] = 0.054$	H-atom parameters constrained
$wR(F^2) = 0.170$	$w = 1/[\sigma^2(F_o^2) + (0.1137P)^2]$
$S = 1.08$	where $P = (F_o^2 + 2F_c^2)/3$
6203 reflections	$(\Delta/\sigma)_{\max} = 0.001$
334 parameters	$\Delta\rho_{\max} = 2.82 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	$\Delta\rho_{\min} = -1.14 \text{ e } \text{\AA}^{-3}$
	Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Sn1	0.347644 (10)	0.11246 (3)	0.186776 (13)	0.04892 (14)
C1	0.37845 (18)	-0.0948 (5)	0.2087 (2)	0.0555 (11)
C2	0.36527 (18)	-0.1935 (5)	0.2426 (2)	0.0593 (11)
H2	0.3448	-0.1670	0.2602	0.071*
C3	0.3825 (2)	-0.3329 (6)	0.2507 (3)	0.0728 (13)
H3	0.3731	-0.3993	0.2724	0.087*
C4	0.4137 (3)	-0.3694 (5)	0.2257 (3)	0.0765 (16)
H4	0.4251	-0.4616	0.2304	0.092*
C5	0.4280 (2)	-0.2738 (7)	0.1948 (3)	0.0810 (15)
H5	0.4495	-0.2999	0.1788	0.097*
C6	0.4110 (2)	-0.1384 (6)	0.1867 (3)	0.0693 (13)
H6	0.4216	-0.0735	0.1657	0.083*
C7	0.27208 (17)	0.1477 (6)	0.1675 (2)	0.0567 (11)
C8	0.2439 (2)	0.0377 (8)	0.1703 (3)	0.0898 (17)
H8	0.2588	-0.0513	0.1823	0.108*
C9	0.1931 (2)	0.0564 (9)	0.1553 (4)	0.0997 (19)
H9	0.1751	-0.0185	0.1588	0.120*
C10	0.1708 (2)	0.1853 (9)	0.1356 (3)	0.094 (2)
H10	0.1365	0.1988	0.1225	0.113*
C11	0.1984 (3)	0.2913 (9)	0.1352 (5)	0.121 (3)
H11	0.1839	0.3807	0.1245	0.145*
C12	0.2486 (3)	0.2729 (8)	0.1503 (4)	0.102 (2)
H12	0.2664	0.3502	0.1485	0.122*
C13	0.37577 (16)	0.2455 (4)	0.1378 (2)	0.0487 (9)

supplementary materials

C14	0.3425 (2)	0.3145 (5)	0.0722 (3)	0.0650 (12)
H14	0.3066	0.2999	0.0481	0.078*
C15	0.3608 (3)	0.4014 (6)	0.0428 (4)	0.0840 (18)
H15	0.3374	0.4478	-0.0002	0.101*
C16	0.4128 (4)	0.4219 (6)	0.0753 (5)	0.096 (2)
H16	0.4253	0.4799	0.0540	0.116*
C17	0.4480 (3)	0.3552 (6)	0.1413 (4)	0.0890 (19)
H17	0.4839	0.3697	0.1645	0.107*
C18	0.42890 (18)	0.2687 (5)	0.1710 (3)	0.0672 (12)
H18	0.4522	0.2243	0.2147	0.081*
C19	0.44734 (17)	0.1592 (6)	0.4717 (3)	0.0637 (12)
C20	0.4307 (2)	0.0166 (7)	0.4709 (4)	0.088 (2)
H20	0.4080	-0.0268	0.4267	0.105*
C21	0.4477 (3)	-0.0562 (8)	0.5337 (5)	0.0917 (18)
H21	0.4369	-0.1483	0.5315	0.110*
C22	0.4809 (2)	0.0069 (7)	0.6004 (4)	0.0861 (16)
H22	0.4915	-0.0423	0.6427	0.103*
C23	0.4981 (2)	0.1400 (6)	0.6043 (3)	0.0698 (13)
H23	0.5211	0.1796	0.6495	0.084*
C24	0.48187 (17)	0.2212 (5)	0.5406 (3)	0.0612 (11)
C25	0.48861 (15)	0.4313 (5)	0.4933 (2)	0.0549 (10)
C26	0.45349 (18)	0.3773 (5)	0.4212 (2)	0.0632 (13)
H26	0.4448	0.4320	0.3812	0.076*
C27	0.43283 (16)	0.2450 (6)	0.4116 (3)	0.0658 (14)
C28	0.3947 (2)	0.1862 (8)	0.3373 (3)	0.087 (2)
C29	0.2256 (3)	-0.1292 (7)	-0.0722 (4)	0.109 (3)
H29A	0.2124	-0.1853	-0.1149	0.163*
H29C	0.2084	-0.0397	-0.0852	0.163*
H29B	0.2190	-0.1759	-0.0393	0.163*
C30	0.3076 (4)	-0.1706 (11)	-0.0725 (6)	0.151 (4)
H30B	0.2818	-0.2201	-0.1154	0.226*
H30C	0.3346	-0.2345	-0.0397	0.226*
H30A	0.3226	-0.0972	-0.0853	0.226*
C31	0.3084 (2)	-0.0349 (5)	0.0229 (3)	0.0668 (12)
H31	0.3437	-0.0186	0.0418	0.080*
N1	0.50173 (14)	0.3517 (4)	0.55044 (19)	0.0541 (8)
N2	0.2823 (2)	-0.1090 (4)	-0.0371 (3)	0.0769 (13)
O1	0.40005 (16)	0.2196 (6)	0.2896 (2)	0.1038 (15)
O2	0.3613 (2)	0.0932 (5)	0.3264 (2)	0.0980 (14)
O3	0.29056 (13)	0.0138 (3)	0.05551 (17)	0.0662 (8)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Sn1	0.03819 (19)	0.0677 (2)	0.03760 (19)	0.00386 (11)	0.01859 (14)	0.00340 (11)
C1	0.044 (2)	0.078 (3)	0.042 (2)	0.0083 (19)	0.0216 (19)	0.0107 (19)
C2	0.056 (3)	0.066 (3)	0.053 (2)	0.006 (2)	0.029 (2)	0.001 (2)
C3	0.062 (3)	0.082 (3)	0.058 (3)	-0.006 (3)	0.023 (2)	-0.002 (3)

C4	0.075 (4)	0.064 (3)	0.075 (4)	0.010 (2)	0.030 (3)	-0.005 (2)
C5	0.071 (3)	0.099 (4)	0.084 (4)	0.021 (3)	0.049 (3)	0.001 (3)
C6	0.062 (3)	0.089 (4)	0.063 (3)	0.016 (2)	0.038 (3)	0.010 (2)
C7	0.042 (2)	0.087 (3)	0.043 (2)	0.004 (2)	0.0235 (18)	-0.002 (2)
C8	0.059 (3)	0.105 (4)	0.110 (5)	0.011 (3)	0.049 (3)	0.014 (4)
C9	0.064 (4)	0.131 (5)	0.112 (5)	-0.008 (4)	0.054 (4)	0.004 (5)
C10	0.053 (3)	0.157 (6)	0.078 (4)	0.025 (4)	0.039 (3)	0.001 (4)
C11	0.081 (4)	0.120 (6)	0.174 (8)	0.037 (4)	0.077 (5)	0.020 (5)
C12	0.070 (4)	0.103 (5)	0.141 (6)	0.023 (3)	0.063 (4)	0.024 (4)
C13	0.049 (2)	0.052 (2)	0.048 (2)	-0.0061 (17)	0.0279 (18)	-0.0119 (17)
C14	0.065 (3)	0.079 (3)	0.054 (3)	-0.006 (2)	0.035 (2)	-0.006 (2)
C15	0.122 (6)	0.070 (3)	0.083 (4)	0.006 (3)	0.070 (4)	0.012 (3)
C16	0.136 (7)	0.064 (3)	0.144 (7)	-0.009 (4)	0.112 (6)	0.003 (4)
C17	0.079 (4)	0.080 (4)	0.133 (6)	-0.027 (3)	0.074 (4)	-0.027 (4)
C18	0.051 (3)	0.075 (3)	0.076 (3)	-0.006 (2)	0.034 (2)	-0.017 (2)
C19	0.037 (2)	0.092 (3)	0.060 (3)	0.001 (2)	0.025 (2)	-0.022 (3)
C20	0.048 (3)	0.097 (4)	0.116 (5)	-0.016 (3)	0.043 (3)	-0.049 (4)
C21	0.077 (4)	0.089 (4)	0.126 (6)	0.007 (3)	0.066 (4)	0.018 (4)
C22	0.071 (4)	0.105 (4)	0.091 (4)	0.007 (3)	0.050 (3)	0.008 (4)
C23	0.065 (3)	0.078 (3)	0.074 (3)	0.004 (2)	0.042 (3)	0.000 (3)
C24	0.042 (2)	0.084 (3)	0.058 (3)	-0.002 (2)	0.028 (2)	-0.015 (2)
C25	0.039 (2)	0.089 (3)	0.0334 (18)	0.0039 (19)	0.0180 (17)	-0.013 (2)
C26	0.042 (2)	0.106 (4)	0.035 (2)	0.005 (2)	0.0172 (19)	-0.016 (2)
C27	0.034 (2)	0.099 (4)	0.059 (3)	-0.009 (2)	0.022 (2)	-0.039 (3)
C28	0.052 (3)	0.156 (6)	0.054 (3)	-0.018 (3)	0.029 (2)	-0.045 (4)
C29	0.091 (5)	0.098 (5)	0.091 (5)	-0.041 (4)	0.020 (4)	-0.013 (3)
C30	0.180 (9)	0.156 (7)	0.193 (10)	-0.066 (7)	0.151 (9)	-0.092 (7)
C31	0.059 (3)	0.073 (3)	0.066 (3)	-0.020 (2)	0.032 (2)	-0.014 (2)
N1	0.0448 (19)	0.074 (2)	0.0385 (17)	0.0026 (17)	0.0192 (15)	-0.0070 (16)
N2	0.086 (3)	0.078 (3)	0.072 (3)	-0.031 (2)	0.047 (3)	-0.023 (2)
O1	0.062 (2)	0.178 (5)	0.048 (2)	-0.017 (3)	0.0146 (18)	-0.008 (2)
O2	0.087 (3)	0.118 (3)	0.071 (3)	-0.031 (2)	0.032 (2)	-0.003 (2)
O3	0.066 (2)	0.072 (2)	0.0561 (18)	-0.0086 (16)	0.0304 (16)	-0.0120 (15)

Geometric parameters (Å, °)

Sn1—C13	2.125 (4)	C17—C18	1.364 (8)
Sn1—C7	2.127 (4)	C17—H17	0.9300
Sn1—C1	2.127 (4)	C18—H18	0.9300
Sn1—O1	2.187 (4)	C19—C27	1.407 (8)
C1—C6	1.384 (7)	C19—C24	1.422 (7)
C1—C2	1.390 (6)	C19—C20	1.447 (8)
C2—C3	1.404 (7)	C20—C21	1.371 (9)
C2—H2	0.9300	C20—H20	0.9300
C3—C4	1.377 (8)	C21—C22	1.387 (10)
C3—H3	0.9300	C21—H21	0.9300
C4—C5	1.344 (8)	C22—C23	1.358 (8)
C4—H4	0.9300	C22—H22	0.9300
C5—C6	1.366 (7)	C23—C24	1.433 (8)

supplementary materials

C5—H5	0.9300	C23—H23	0.9300
C6—H6	0.9300	C24—N1	1.349 (6)
C7—C12	1.337 (8)	C25—N1	1.331 (6)
C7—C8	1.378 (8)	C25—C25 ⁱ	1.436 (10)
C8—C9	1.406 (8)	C25—C26	1.443 (6)
C8—H8	0.9300	C26—C27	1.373 (7)
C9—C10	1.356 (11)	C26—H26	0.9300
C9—H9	0.9300	C27—C28	1.507 (6)
C10—C11	1.318 (10)	C28—O1	1.190 (7)
C10—H10	0.9300	C28—O2	1.270 (8)
C11—C12	1.385 (9)	C29—N2	1.471 (9)
C11—H11	0.9300	C29—H29A	0.9600
C12—H12	0.9300	C29—H29C	0.9600
C13—C18	1.383 (6)	C29—H29B	0.9600
C13—C14	1.397 (7)	C30—N2	1.482 (9)
C14—C15	1.345 (8)	C30—H30B	0.9600
C14—H14	0.9300	C30—H30C	0.9600
C15—C16	1.351 (11)	C30—H30A	0.9600
C15—H15	0.9300	C31—O3	1.206 (6)
C16—C17	1.403 (11)	C31—N2	1.316 (6)
C16—H16	0.9300	C31—H31	0.9300
C13—Sn1—C7	119.75 (18)	C18—C17—H17	120.4
C13—Sn1—C1	114.85 (16)	C16—C17—H17	120.4
C7—Sn1—C1	119.16 (19)	C17—C18—C13	121.7 (6)
C13—Sn1—O1	88.78 (16)	C17—C18—H18	119.2
C7—Sn1—O1	103.41 (17)	C13—C18—H18	119.2
C1—Sn1—O1	102.48 (19)	C27—C19—C24	115.7 (5)
C6—C1—C2	117.2 (4)	C27—C19—C20	127.4 (5)
C6—C1—Sn1	120.7 (4)	C24—C19—C20	117.0 (5)
C2—C1—Sn1	122.0 (3)	C21—C20—C19	121.8 (6)
C1—C2—C3	120.7 (5)	C21—C20—H20	119.1
C1—C2—H2	119.7	C19—C20—H20	119.1
C3—C2—H2	119.7	C20—C21—C22	120.3 (6)
C4—C3—C2	118.8 (5)	C20—C21—H21	119.9
C4—C3—H3	120.6	C22—C21—H21	119.9
C2—C3—H3	120.6	C23—C22—C21	120.4 (6)
C5—C4—C3	121.0 (5)	C23—C22—H22	119.8
C5—C4—H4	119.5	C21—C22—H22	119.8
C3—C4—H4	119.5	C22—C23—C24	121.8 (6)
C4—C5—C6	120.2 (5)	C22—C23—H23	119.1
C4—C5—H5	119.9	C24—C23—H23	119.1
C6—C5—H5	119.9	N1—C24—C19	124.1 (5)
C5—C6—C1	122.1 (5)	N1—C24—C23	117.2 (4)
C5—C6—H6	119.0	C19—C24—C23	118.7 (5)
C1—C6—H6	119.0	N1—C25—C25 ⁱ	117.6 (4)
C12—C7—C8	116.2 (5)	N1—C25—C26	119.9 (5)
C12—C7—Sn1	123.5 (4)	C25 ⁱ —C25—C26	122.5 (5)
C8—C7—Sn1	120.2 (4)	C27—C26—C25	120.0 (5)

C7—C8—C9	121.7 (6)	C27—C26—H26	120.0
C7—C8—H8	119.1	C25—C26—H26	120.0
C9—C8—H8	119.1	C26—C27—C19	120.5 (4)
C10—C9—C8	119.0 (7)	C26—C27—C28	121.7 (6)
C10—C9—H9	120.5	C19—C27—C28	117.8 (5)
C8—C9—H9	120.5	O1—C28—O2	120.1 (5)
C11—C10—C9	119.2 (6)	O1—C28—C27	117.7 (6)
C11—C10—H10	120.4	O2—C28—C27	121.8 (6)
C9—C10—H10	120.4	N2—C29—H29A	109.5
C10—C11—C12	121.7 (7)	N2—C29—H29C	109.5
C10—C11—H11	119.2	H29A—C29—H29C	109.5
C12—C11—H11	119.2	N2—C29—H29B	109.5
C7—C12—C11	122.1 (7)	H29A—C29—H29B	109.5
C7—C12—H12	118.9	H29C—C29—H29B	109.5
C11—C12—H12	118.9	N2—C30—H30B	109.5
C18—C13—C14	116.8 (4)	N2—C30—H30C	109.5
C18—C13—Sn1	120.4 (4)	H30B—C30—H30C	109.5
C14—C13—Sn1	122.7 (3)	N2—C30—H30A	109.5
C15—C14—C13	122.1 (5)	H30B—C30—H30A	109.5
C15—C14—H14	119.0	H30C—C30—H30A	109.5
C13—C14—H14	119.0	O3—C31—N2	125.7 (5)
C14—C15—C16	120.5 (6)	O3—C31—H31	117.2
C14—C15—H15	119.7	N2—C31—H31	117.2
C16—C15—H15	119.7	C25—N1—C24	119.8 (4)
C15—C16—C17	119.7 (5)	C31—N2—C29	119.4 (5)
C15—C16—H16	120.2	C31—N2—C30	122.6 (6)
C17—C16—H16	120.2	C29—N2—C30	118.0 (6)
C18—C17—C16	119.1 (6)	C28—O1—Sn1	114.5 (4)
C13—Sn1—C1—C6	-4.8 (5)	C15—C16—C17—C18	1.0 (10)
C7—Sn1—C1—C6	147.4 (4)	C16—C17—C18—C13	-0.2 (8)
O1—Sn1—C1—C6	-99.3 (4)	C14—C13—C18—C17	0.2 (7)
C13—Sn1—C1—C2	178.0 (3)	Sn1—C13—C18—C17	-179.1 (4)
C7—Sn1—C1—C2	-29.8 (4)	C27—C19—C20—C21	-179.9 (5)
O1—Sn1—C1—C2	83.5 (4)	C24—C19—C20—C21	-0.5 (7)
C6—C1—C2—C3	-2.9 (7)	C19—C20—C21—C22	1.2 (8)
Sn1—C1—C2—C3	174.3 (4)	C20—C21—C22—C23	-1.8 (9)
C1—C2—C3—C4	1.4 (7)	C21—C22—C23—C24	1.6 (8)
C2—C3—C4—C5	0.6 (8)	C27—C19—C24—N1	-2.1 (6)
C3—C4—C5—C6	-0.9 (9)	C20—C19—C24—N1	178.5 (4)
C4—C5—C6—C1	-0.8 (9)	C27—C19—C24—C23	179.8 (4)
C2—C1—C6—C5	2.7 (8)	C20—C19—C24—C23	0.3 (6)
Sn1—C1—C6—C5	-174.6 (4)	C22—C23—C24—N1	-179.2 (5)
C13—Sn1—C7—C12	-26.3 (6)	C22—C23—C24—C19	-0.9 (7)
C1—Sn1—C7—C12	-177.2 (5)	N1—C25—C26—C27	1.3 (6)
O1—Sn1—C7—C12	70.0 (5)	C25 ⁱ —C25—C26—C27	-180.0 (5)
C13—Sn1—C7—C8	151.7 (4)	C25—C26—C27—C19	-2.6 (6)
C1—Sn1—C7—C8	0.9 (5)	C25—C26—C27—C28	178.1 (4)
O1—Sn1—C7—C8	-111.9 (4)	C24—C19—C27—C26	2.9 (6)

supplementary materials

C12—C7—C8—C9	0.5 (9)	C20—C19—C27—C26	-177.7 (4)
Sn1—C7—C8—C9	-177.7 (5)	C24—C19—C27—C28	-177.8 (4)
C7—C8—C9—C10	2.0 (10)	C20—C19—C27—C28	1.6 (7)
C8—C9—C10—C11	-4.1 (11)	C26—C27—C28—O1	33.8 (9)
C9—C10—C11—C12	3.8 (13)	C19—C27—C28—O1	-145.5 (6)
C8—C7—C12—C11	-0.8 (11)	C26—C27—C28—O2	-153.2 (6)
Sn1—C7—C12—C11	177.3 (7)	C19—C27—C28—O2	27.5 (8)
C10—C11—C12—C7	-1.3 (14)	C25 ⁱ —C25—N1—C24	-179.2 (4)
C7—Sn1—C13—C18	151.2 (3)	C26—C25—N1—C24	-0.4 (6)
C1—Sn1—C13—C18	-56.8 (4)	C19—C24—N1—C25	0.9 (6)
O1—Sn1—C13—C18	46.4 (4)	C23—C24—N1—C25	179.0 (4)
C7—Sn1—C13—C14	-28.0 (4)	O3—C31—N2—C29	4.4 (8)
C1—Sn1—C13—C14	124.0 (4)	O3—C31—N2—C30	-177.3 (7)
O1—Sn1—C13—C14	-132.8 (4)	O2—C28—O1—Sn1	1.2 (9)
C18—C13—C14—C15	-1.0 (7)	C27—C28—O1—Sn1	174.3 (4)
Sn1—C13—C14—C15	178.2 (4)	C13—Sn1—O1—C28	173.5 (5)
C13—C14—C15—C16	1.9 (9)	C7—Sn1—O1—C28	53.1 (6)
C14—C15—C16—C17	-1.9 (10)	C1—Sn1—O1—C28	-71.3 (5)

Symmetry codes: (i) $-x+1, -y+1, -z+1$.

Fig. 1

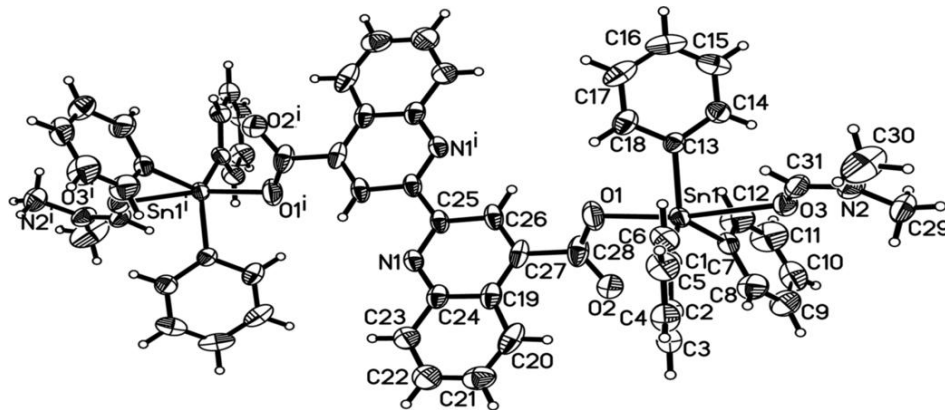


Fig. 2

